

This article was downloaded by:

On: 18 January 2011

Access details: *Access Details: Free Access*

Publisher *Taylor & Francis*

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



International Journal of Polymeric Materials

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t713647664>

Effects on Impact and Tensile Behavior of Transparent Polymer Materials by the Addition of Anti-Biofouling Agents

Chad Booth^a; Phil Wheeler^a; Jessie Hancock^a; Ray Ximenes^a; Donald E. Patterson^b

^a Department of Chemistry and Biochemistry, Texas State University, San Marcos, Texas ^b Nanohmics, Inc., Austin, Texas

To cite this Article Booth, Chad , Wheeler, Phil , Hancock, Jessie , Ximenes, Ray and Patterson, Donald E.(2008) 'Effects on Impact and Tensile Behavior of Transparent Polymer Materials by the Addition of Anti-Biofouling Agents', International Journal of Polymeric Materials, 57: 5, 452 – 462

To link to this Article: DOI: 10.1080/00914030701729628

URL: <http://dx.doi.org/10.1080/00914030701729628>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.informaworld.com/terms-and-conditions-of-access.pdf>

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

Effects on Impact and Tensile Behavior of Transparent Polymer Materials by the Addition of Anti-Biofouling Agents

Chad Booth
Phil Wheeler
Jessie Hancock
Ray Ximenes

Department of Chemistry and Biochemistry, Texas State University,
San Marcos, Texas

Donald E. Patterson
Nanohmics, Inc., Austin, Texas

This paper reports the effects on both the impact and tensile behavior, impact resistance, modulus, strain at break, and peak load for a series of polymers with anti-biofouling agents added. The materials include PMMA, bisphenol-A polycarbonate, and a co-polyterephthalate derived from 2,2,4,4-tetramethyl-1,3-cyclobutanediol and 1,3-propanediol (CBDO). The anti-biofouling agents studied were the algaecides Irgarol 1051 and Diuron. The anti-biofouling agents were not simply used as a coating but were compounded into pelletized material and then compression molded into test plaques. The results show that PMMA is the only polymeric material studied that does not show a decrease in impact resistance with an increasing quantity of the anti-biofouling agent. With respect to modulus, strain at break, and peak load, both PMMA and CBDO showed little dependence on the amount or type of anti-biofouling additive.

Keywords: anti-biofouling, CBDO, impact resistance

INTRODUCTION

Solar panels and bio-optical sensors play a significant and growing role in a number of applications that are of importance to many

Received 13 September 2007; in final form 20 September 2007.

The authors would like to express their gratitude to the National Oceanographic and Atmospheric Administration (NOAA) for funding through a Phase II SBIR, Contract #DG133R06CN0202.

Address correspondence to Chad Booth, Department of Chemistry and Biochemistry, Texas State University, San Marcos, Texas, 78666. E-mail: cb31@txstate.edu

organizations. Many of these instruments (solar panels, spectrophotometers) require a high transmission of radiation (UV, visible, and infrared light) into the device for it to work properly. Typically, these devices have a coverglass affixed over the active part of the device to protect the sensor or panel from harsh ambient conditions. A problem that presently exists with both sensors and solar panels is biological fouling of the coverglasses on these instruments. Over time in an ocean environment, any number of both plants and animals (algae, seaweed, barnacles, tube worms) will attach themselves to the coverglass, ultimately obscuring the glass and rendering the instrument useless. As such, an anti-fouling mechanism is needed for these instruments that is inexpensive, long-lasting, and environmentally friendly.

Two things must be considered when examining optical coverglass materials. These include the optical behavior of the material, percent transmission, UV stability, and the mechanical properties. Polymers such as bisphenol-A polycarbonate (PC) provide excellent impact resistance [1] but do suffer from poor resistance to UV [2] with extended exposure. On the other hand, materials such as poly(methylmethacrylate) (PMMA) display excellent UV stability [3] but suffer in that their impact resistance is significantly lower than that of PC. Over the years there has been a moderate industrial interest in a co-polyterephthalate composed of 2,2,4,4-tetramethyl-1,3-cyclobutanediol (CBDO) and 1,3-propanediol (PDO) (Figure 1) [4–8]. While this material has sparked moderate interest from industry, it has only recently been examined from the academic side. This material displays enhanced UV stability [9], when compared to PC, as well as enhanced impact resistance [10], when compared to both PC and PMMA. This material is a co-polyterephthalate synthesized using 2,2,4,4-tetramethyl-1,3-cyclobutanediol (CBDO) and 1,3-propanediol (PDO) (Figure 1). Notched Izod impact values for the CBDO-PDO co-polyterephthalate have been reported as high as 1070 J/m [9], depending on the percent incorporation of the monomers, and ballistic testing shows the material to be as much as 35% better than PC.

The problem of developing an anti-biofouling optical coverglass material is further complicated because a small molecule additive must be compounded into the material. Polymers such as polycarbonate

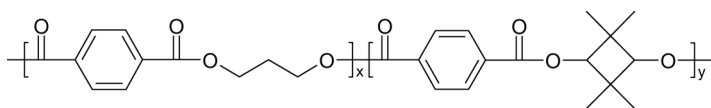


FIGURE 1 CBDO-PDO co-polyterephthalate.

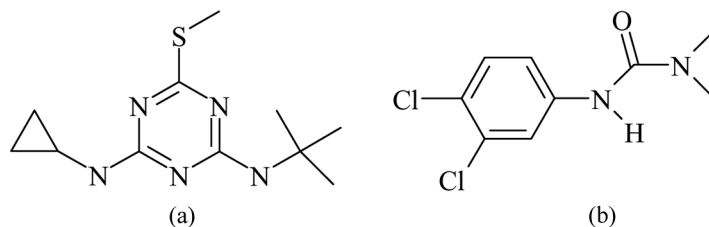


FIGURE 2 Structure of Irgarol[®] 1051 (a) and Diuron (b).

are notorious for showing a decrease in mechanical properties with the introduction of additives [11]. This diminution of mechanical properties is typically less of a problem in vinyl systems such as PMMA, and in many cases an increase in the mechanical properties is observed [12]. Recent studies reported for the CBDO-PDO co-polyterephthalate show that the addition of additives, in this case organoclay nanoparticles, contributes to an increase in stiffness and elongation at break [13]. There is a reduction in impact resistance, but that corresponds to the amount of additive utilized.

This study will address the effects on both the impact and tensile behavior of the addition of two anti-biofouling algacides, N-(tert-butyl)-N'-cyclopropyl-6-(methylthio)-1,3,5-triazine-2,4-diamine (Irgarol[®] 1051) and 3-(3,4-dichlorophenyl)-1,1-dimethylurea (Diuron) (Figure 2), to three optically transparent polymeric materials: PMMA, PC, and the CBDO-PDO co-polyterephthalate. The effects on the materials' impact and tensile properties will be discussed. The effects on the materials' optical properties will be reported in a subsequent paper.

EXPERIMENTAL

Chemicals

Both the PMMA and polycarbonate were purchased in resin form from TDL Plastics (Corpus Christi, Texas). The polycarbonate utilized was Wonderlite[®] PC-110 while the PMMA utilized was a Continental Acrylics CA-1000. The CBDO used was donated by Eastman Chemical Company and has a *cis/trans* ratio of 46/54 with a M_w of 55,000. Two anti fouling agents were examined. The first was Irgarol[®] 1051 (Figure 2). The Irgarol[®] 1051 was donated by Ciba Chemical Company. The second anti biofouling agent was Diuron (Figure 2). The Diuron was purchased from Sigma-Aldrich Chemical Company. All chemicals were used as received after thorough drying.

Characterization

The concentration of the additives in each polymer sample was verified using the Beer-Lambert Law relationship (Equation (1)):

$$A = \alpha l c \quad (1)$$

where A is absorbance, α is the absorption coefficient (L/mol cm), l is the cell path length (cm), and c is the solution concentration (mol/L). UV spectroscopy was carried out utilizing an Ocean Optics S2000 UV-Visible spectrometer with 600 grooves/mm and a blaze wavelength of 300 nm. A deuterium light source was used for the ultraviolet absorption studies. A series of calibration curves was constructed resulting in absorption coefficients of 17086 L/mol cm for Diuron at 254 nm with $R^2 = 0.9987$ and 8697 L/mol cm for Irgarol at 251 nm with $R^2 = 0.9946$.

Several compounded sample pellets were dissolved in CH_2Cl_2 for testing. The targeted values for the compounded Irgarol[®] 1051 samples were 0.25 and 0.5% by weight of Irgarol to polymer while the targeted values for the Diuron compounds were 2 and 4% by weight. The UV concentration studies resulted in actual Irgarol weight fractions of 0.39 and 0.70 and actual Diuron weight fractions of 2.10 and 4.23.

Thermogravimetric analysis (TGA) was obtained using a TA Instruments Q50 TGA. Scans were performed using 5–10 mg samples at a rate of 20°C per min from ambient to 800°C. Decomposition temperatures are reported as the value at which the sample displays a 10% loss in mass.

Differential scanning calorimetry (DSC) was conducted using a TA Instruments Q200 DSC. Scans were obtained using 5 mg samples at a heating/cooling rate of 10°C per min. All scans were conducted under an inert atmosphere. Reported data was collected on the first cooling and second heating cycles.

Modulus data was collected using a MTS/Sintech Model 1D Tensile Tester equipped with a Pentium data station. Samples were standardized and tested according to ASTM D628.

Impact resistance was obtained using a Gardner Impact Tester. This system is a “dart style” tester which utilizes a hammer of specific mass dropped from a calibrated distance. For this study a 227 g hammer was used for the PMMA samples while a 3632 g hammer was used for the PC and CBDO. Plaques utilized for this test were 2.5 cm × 2.5 cm × 0.16 cm.

Processing

The polymer resin was dried using a Napco E series vacuum oven equipped with a Welch 1402 mechanical vacuum pump. The polymer

resins were dried for 48 h at 70°C under vacuum. Processing of the polymer resins with the anti-biofouling agents was accomplished using a HAAKE Rheodrive 500 connected to a HAAKE Rheomax CTW 100 twin screw extruder. PMMA was processed using a temperature range of 108–210°C, PC was processed using a temperature range of 180–200°C, and the CBDO was processed using a temperature range from 150–165°C. Each of the three polymers examined was compounded with 100% and 50% of the manufacturers' suggested quantity of anti-biofouling agent while performing a 2-factorial designed experiment. For the Irgarol[®] 1051, the manufacturers' suggested incorporation is 0.5% (w/w) and for the Diuron, the suggested percent incorporation is 4% (w/w). The premix formulations were compounded by dry blending the individual components and melt-mixing them in the HAAKE twin screw extruder. The extruder was fitted with a monofilament die head, and upon extrusion the compounded polymers were pulled through a water trough cooled using dry ice and then finally into a Haake PP1 Postex Pelletizer. The pelletized compounded polymer resin was then dried at 70°C for 24 h under vacuum.

Test plaques were made using compression molding on a Carver Laboratory Press. Sample plaques were formed using aluminum plates and aluminum molds with dimensions of 18 cm × 18 cm × 0.16 cm. The press was preheated to the predetermined temperature suitable for molding of the polymer composites. The molding temperature for PMMA and polycarbonate was determined to be 204°C, while the molding temperature for CBDO was 163°C. The aluminum molds were sprayed with Universal II mold release to prevent the compound from adhering to the molds. To form the plaques, 60 g of composite resin was placed into the heated mold and left for 5 min. The top mold plate was then placed on the mold and the press was raised to close the mold. The closed mold was allowed to equilibrate for 5 min, after which pressure was slowly increased until a pressure of 1.38×10^8 Pa was reached. The mold was then removed from the press and allowed to cool. The polymer plaque was removed from the mold and then cut and shaped to the measurements specified by the particular test to be performed.

RESULTS AND DISCUSSION

TGA and DSC studies show that all materials examined are amorphous. Additionally, no effect on the thermal decomposition temperature of the materials was observed with the introduction of the additives. The measured T_d for the PMMA was 315°C, the CBDO-PDO co-polyterephthalate was 370°C, and the PC was 415°C.

Several mechanical tests were performed in order to evaluate the effect of adding the anti-biofouling agents to the optical plastics. Impact resistance was evaluated using dart impact studies. Both the CBDO co-polyterephthalate and the bisphenol-A polycarbonate were evaluated using the standard 3632 g hammer, while the PMMA samples were evaluated using a 227 g hammer. The results from impact testing are shown in Table 1.

The data show that bisphenol-A polycarbonate and, to a lesser extent, the CBDO co-polyterephthalate display a decrease in impact strength with increasing anti-biofouling additive. The polycarbonate shows a large decrease with the introduction of the Diuron. What is not clear is whether this is due specifically to the Diuron or simply the amount of additive being introduced. Bisphenol-A polycarbonate is notorious for being sensitive to additives and as much as 4 wt% of the Diuron was used. As discussed earlier, the incorporated weight percentages for both Diuron and Irgarol[®] 1051 were based on the manufacturers' suggested incorporation values and a 2-factorial designed experiment that was run concurrently. In this study, the Diuron was incorporated at 2.1 and 4.2 wt% while the Irgarol[®] 1051 was incorporated at 0.39 and 0.70 wt%.

With respect to the PMMA samples, there is, statistically, no difference in the impact resistance of the neat PMMA samples when compared to the PMMA samples compounded with the anti-biofouling agents.

TABLE 1 Impact Resistance

Material	Plaque 1 (J/m)	Plaque 2 (J/m)	Plaque 3 (J/m)	Plaque 4 (J/m)	Plaque 5 (J/m)	Average (J/m)
PMMA						
Neat	4.45	4.45	4.45	4.45	4.45	4.45
.25% Irgarol	4.45	4.45	4.45	4.45	4.45	4.45
.50% Irgarol	4.45	4.45	6.67	4.45	N/A	5.00
2% Diuron	4.45	6.67	4.45	4.45	N/A	5.00
4% Diuron	4.45	4.45	4.45	2.22	N/A	3.89
CBDO						
Neat	1031	961	996	961	925	975
.25% Irgarol	890	961	925	925	925	925
.50% Irgarol	890	818	854	890	N/A	863
2% Diuron	854	747	818	854	N/A	818
4% Diuron	818	712	605	427	N/A	641
Bis A-PC						
Neat	925	890	925	925	N/A	916
.25% Irgarol	712	783	854	N/A	N/A	783
.50% Irgarol	783	712	605	N/A	N/A	700
2% Diuron	2.22	2.22	4.45	4.45	N/A	3.34
% Diuron	4.45	4.45	4.45	6.67	N/A	5.00

In addition to the impact values, DMA was run to obtain tensile data, peak load, peak stress, percent strain at break, and modulus. The values are shown in Table 2 and reflect the neat, 0.39 and 0.70% Irgarol[®] 1051, and neat, 2.1 and 4.2% Diuron for all three polymer samples.

Impact resistance for the neat samples is in good agreement with published values. The CBDO-PDO co-polyterephthalates displayed a 6% higher impact resistance when compared to bisphenol-A polycarbonate. Earlier reports [10], which utilized injection molding rather than compression molding, show that the CBDO-PDO co-polyterephthalates are capable of up to a 35% increase in impact resistance over bisphenol-A polycarbonate. We attribute this difference to the use of compression molding rather than injection molding.

The PMMA shows no difference in impact resistance with either of the anti-biofouling agents at any loading. This result is not unexpected; PMMA has a good reputation of accepting and handling additives without loss of thermal or mechanical properties [14]. With respect to the tensile properties the PMMA displays only a moderate decrease in peak load, peak stress, percent strain at break, and modulus. In most cases, the 0.39% loading of Irgarol[®] actually increases the properties. This increase is most likely due to the reinforcement effects of the additives as has been shown with other PMMA additives in other studies [15–16]. The general findings show that at low loadings the mechanical properties are increased. This finding is attributed to the additive acting to stiffen regions of the amorphous polymer and create regions which are self-supporting—similar to the behavior of semi-crystalline materials. At higher loadings, the additives simply limit the polymer chain motion to the point that the material becomes more brittle and a loss of mechanical reinforcement is observed.

The CBDO-PDO co-polyterephthalate does display an almost linear decrease in impact resistance with increasing additive. The overall impact resistance falls 34% with the addition of 4.2 wt% Diuron; but at 641 J/m, this value is still high enough to qualify as a high-impact resistance material. As with the PMMA, this is not unexpected; the CBDO-PDO co-polyterephthalate has been reported to display a loss in impact resistance with increased additive loading [13]. Other work has been carried out using organo-clay nanoparticles as the additives, and the reasons reported are that the clay particles are tactoidal in nature and therefore act to stiffen the material at low loadings; however, with increased percent loadings, the nanoparticles impart brittleness to the polymer system.

TABLE 2 Mechanical Values

Material	Peak load (N)	Peak stress (N/mm ²)	% Strain at break	Modulus (N/mm ²)	
PMMA	Neat	Run 1 1170	708	4.2	21348
		Run 2 1557	956	5.9	23142
		Run 3 1338	788	5.5	19728
		Average 1355	817	5.2	21406
	.25% Irgarol	Run 1 1366	1150	6.4	27080
		Run 2 1069	781	4.1	23379
		Run 3 1328	1001	5.5	26454
		Average 1255	977	5.3	25637
	.50% Irgarol	Run 1 1019	520	3.0	19857
		Run 2 1542	619	5.8	14918
	Run 3 1260	596	4.1	18091	
	Average 1273	578	4.3	17622	
	Run 1 1200	978	4.4	26938	
	Run 2 1287	769	4.9	20006	
	Run 3 1457	847	5.6	19793	
	Average 1315	865	5.0	22246	
	Run 1 927	546	3.6	17575	
	Run 2 1255	1023	5.5	25991	
	Run 3 1144	665	3.9	20020	
	Average 1109	745	4.3	21195	
	Run 1 1566	14	101.5	443	
	Run 2 1568	14	N/A	440	
	Run 3 1522	13	172.9	432	
	Average 1552	13.7	137.2	439	
CBDO	.25% Irgarol	Run 1 1652	16	171.5	512
		Run 2 1684	16	145.4	525
		Run 3 1710	16	159.1	516
		Average 1682	16	158.7	517

(Continued)

TABLE 2 Continued

Material	Peak load (N)	Peak stress (N/mm ²)	% Strain at break	Modulus (N/mm ²)
.50% Irgarol	Run 1	1739	83.4	486
	Run 2	1745	N/A	497
	Run 3	1656	154.8	488
	Average	1713	119.1	490
2% Diuron	Run 1	1836	131	479
	Run 2	1768	152	511
	Run 3	1793	102	515
	Average	1799	128	501
4% Diuron	Run 1	1795	163	506
	Run 2	1784	64	485
	Run 3	1792	136	500
	Average	1790	121	497
Neat	Run 1	2532	10.3	7174
	Run 2	2446	11.1	7218
	Run 3	2474	10.3	7140
	Average	2484	10.6	7177
.25% Irgarol	Run 1	2861	6.3	6479
	Run 2	1946	3.4	6792
	Run 3	2105	206	4478
	Average	2304	4.1	5916
.50% Irgarol	Run 1	2300	13.1	4051
	Run 2	2192	5.1	8700
	Run 3	2609	8.9	8366
	Average	2367	9.0	7029
2% Diuron	Run 1	1014	1.4	8475
	Run 2	813	1.1	8351
	Run 3	1124	1.7	7288
	Average	984	1.4	8038
4% Diuron	Run 1	709	0.9	8052
	Run 2	524	0.7	8977
	Run 3	1254	1.8	7850
	Average	829	1.1	8293

Bis A-PC

The effects on the tensile properties of the CBDO-PDO co-polyterephthalate with increasing anti-biofouling additive are unlike the other samples investigated. For all tensile values examined, peak load, peak stress, percent strain at break, and modulus, the values showed an increase with the lower loading of each additive and then a subsequent decrease at higher loadings of each additive. Work reported using organo-clay nanocomposites of the CBDO-PDO co-polyterephthalate showed this same trend [13]. The material showed an increase in tensile values with up to 10 wt% loading of the organo-clay particles. This increase in tensile values was attributed to orientation effects created not only by injection molding but also the large aspect ratio of the organo-clay particles. That does not seem to fit the model here in that the additives used in this study are small organic molecules and not large aspect ratio organo-silicates. More likely, the explanation is related to the self-orientation of the polymer chains themselves. For flexible polymers, such as PMMA, injection molding imparts more order than compression molding. But for polymers that contain rigid units connected by flexible units, such as the CBDO-PDO co-polyterephthalate, there is an intrinsic amount of ordering which occurs with any processing; whether it be compression or injection molding. This ordering is likely the cause of the increase in tensile properties.

The bisphenol-A polycarbonate shows the most dramatic effects with increasing additive. The neat PC samples have an average impact resistance of 916 J/m. The samples with 0.39 and 0.70% Irgarol[®] added fall to 783 and 700 J/m, respectively. With the incorporation of the Diuron, the impact resistance falls to values lower than those reported for the PMMA samples. This dramatic decrease in the observed impact resistance is most commonly associated with a decrease in the motion of the phenylene units of the bisphenol-A group [17]. The introduction of small organic molecules which contain polar functionality limits the allowed rotation of the phenylene units via electrostatic interactions.

CONCLUSIONS

For the PC samples, all tensile properties, except modulus, display a decrease with increasing percent additive. As discussed above, the increase in stiffness of the material is best explained by examining the decrease in phenylene unit rotation [17]. The decrease in polymer chain (specifically the phenylene units) rotation will without doubt add to the stiffness of the material.

REFERENCES

- [1] Karbach, A., Drechsler, D., Schultz, C. L., Wollborn, U., Moethrath, M., and Erkelenz, M. (2005). In *Advances in Polycarbonates, Symposium Series*. D. J. Brunelle and M. R. Korn, Eds., American Chemical Society, Washington, DC, **898**, pp. 112.
- [2] Suriano, J. A., Siclovan, T. M., Pickett, J. E., Brunelle, D. J., O'Neil, G. A., and Zhou, H., *ACS Polymer Preprints* **44**, 674 (2003).
- [3] Wunderlich, W. (1989). In *Polymer Handbook*. J. Brandrup and E. H. Immergut, Eds., John Wiley & Sons, New York, **3**, pp. V77–80.
- [4] Sumner, C. E., Jr., Gustafson, B. L., and Knight, J. R. (Eastman Kodak) U.S. Patent 5,169,994, Dec 8, 1992.
- [5] Sumner, C. E., Jr., Gustafson, B. L., and Knight, J. R. (Eastman Kodak) U.S. Patent 5,258,556, Nov 2, 1993.
- [6] Elam, E. U. and Hasek, R. H. (Eastman Kodak) U.S. Patent 3,190,928, June 22, 1965.
- [7] Kelsey, D. R. (Shell) U.S. Patent 5,705,575, Jan 6, 1998.
- [8] Elam, E. U., Martin, J. C., and Gilkey, R. (Kodak) U.S. Patent 3,313,777, April 11, 1967.
- [9] Kelsey, D. R., Scardino, B. M., Grebowicz, J. S., and Chuah, H. H., *Macromolecules* **33**, 5810 (2000).
- [10] Booth, C. J., Kindinger, M., McKenzie, H. R., Hancock, J., Bray, A. V., and Beall, G. W., *Polymer* **47**, 6398 (2006).
- [11] Lee, S. S., Cho, K., and Kim, J., *Journal of Polymer Science; Part B: Polymer Physics* **39**, 2635 (2001).
- [12] Kotha, S. P., Li, C., McGinn, P., Schmid, S. R., and Mason, J. J., *Journal of Materials Science: Materials in Medicine* **17**, 1403 (2006).
- [13] Beall, G. W., Powell, C. E., Hancock, J., Kindinger, M., McKenzie, H. R., Bray, A. V., and Booth, C. J., *Applied Clay Science* **37**, 295 (2007).
- [14] Fu, J. and Naguib, H. E., *Journal of Cellular Plastics* **42**, 325 (2006).
- [15] Penn, L. S., Maillot, J. C., Anthony, J., Payne, M. M., Swiss, B. D., Mallory, F. B., Liu, K., and Miller, R. A., *Proceeding of the Annual Meeting of the Adhesion Society* **27**, 99 (2004).
- [16] Classen, M. S., Schmidt, W. H., and Wendorff, J. H., *Polymers for Advanced Technologies* **1**, 143 (1990).
- [17] Wehrle, M., Hellmann, G. P., and Spiess, H. W., *Colloid and Polymer Science* **265**, 815 (2005).